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HARNESS, DICKEY & PIERCE, P.L.C. P.O. BOX 828 BLOOMFIELD HILLS, MI 48303			RIDLEY, BASIA ANNA	
			ART UNIT	PAPER NUMBER
			1764	

DATE MAILED: 09/21/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

Office Action Summary	Application No. 09/780,184	Applicant(s) GITTLEMAN ET AL.	
	Examiner Basia Ridley 	Art Unit 1764	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 17 June 2004.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 2-11 and 14-47 is/are pending in the application.
- 4a) Of the above claim(s) 19,24 and 31-45 is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 2-11, 14-18, 20-23, 25-30, 46 and 47 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
 2. ☐ Certified copies of the priority documents have been received in Application No. _____.
 3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- | | |
|---|---|
| 1) <input type="checkbox"/> Notice of References Cited (PTO-892) | 4) <input type="checkbox"/> Interview Summary (PTO-413)
Paper No(s)/Mail Date. _____ |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948) | 5) <input type="checkbox"/> Notice of Informal Patent Application (PTO-152) |
| 3) <input checked="" type="checkbox"/> Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08)
Paper No(s)/Mail Date <u>032904</u> . | 6) <input type="checkbox"/> Other: _____ |

DETAILED ACTION

Claim Rejections - 35 USC § 112

1. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.
2. Claim(s) 2-11, 14-18, 20-23, 25-30 and 46-74 are rejected under 35 U.S.C. 112, first paragraph, because the specification, while being enabling for a system comprising a water-gas shift reactor wherein a layer of carbon monoxide adsorbent may be placed in front of the water gas shift catalyst or the two materials may be combined into a single layer (page 11, lines 34-36), does not reasonably provide enablement for a second reactor which is a water gas shift reactor, wherein the water gas shift reactor includes an adsorbent adapted to adsorb carbon monoxide, as recited in claims 20 and 46. The specification does not enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make and operate a system commensurate in scope with these claims. The broad claim language encompasses all reactors which comprise both, a shift catalyst and carbon monoxide adsorbent, while the specification only provides enablement for two specific embodiments of said reactors. The specification does not enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make and operate a system, wherein, for example the adsorbent zone included in the shift reactor is located downstream from the water gas shift zone.

Claim Rejections - 35 USC § 103

3. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

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4. Claims 7, 29 and 46 are rejected under 35 U.S.C. 103(a) as being unpatentable over Bellows et al. (USP 5,604,047).

Regarding claim(s) 46, Bellows et al. disclose(s) system comprising:

- a first reactor which produces a hydrogen-containing stream gas stream comprising CO (C2/L28-36);
- a second reactor, which is a water-gas shift reactor disposed downstream of said first reactor (C2/L28-36).

While the reference discloses a first adsorbent (21) adapted to adsorb the carbon monoxide (Fig. 1), it does not explicitly disclose said first adsorbent located in the second reactor. As the instant specification is silent to unexpected results, it would have been prima facie obvious for one having ordinary skill in the art to combine the second reactor and the third reactor (20) having the first adsorbent into one vessel, with the reasonable expectation of eliminating the need for a two separate vessels. One of ordinary skill in the art would have been motivated to forgo use of separate vessels, by combining their function and benefit, where doing so is technically feasible and would reduce cost. See *In re Thompson*, 545 F.2d 1290, 1229, 188 USPQ 365, 367 (CCPA 1976). Therefore, one of ordinary skill in the art would have expected that the combination of water-shift reactor and the adsorbent vessel would avoid costs associated with the installation and operation of two vessels. *In re Thompson*, 545 F.2d 1290, 1294, 192 USPQ 275, 277 (CCPA 1976); *In re Clinton*, 527 F.2d 1226, 1229, 188 USPQ 365, 367 (CCPA 1976). Further, the examiner notes that said modification would merely involve making elements integral, and it has been held that making elements integral is generally recognized as being within the level of ordinary skill in the art. *In re Larson*, 340 F.2d 965, 968, 144 USPQ 347, 349 (CCPA 1965).

Regarding claims 7 and 29 Bellows et al. discloses all of the claim limitations as set forth above. Additionally the reference discloses the system wherein:

- the system is a fuel cell system (C1/L14-45); and wherein
- the preferential oxidizer is eliminated from the system (Fig. 1).

5. Claim 14 is rejected under 35 U.S.C. 103(a) as being unpatentable over Bellows et al. (USP 5,604,047) in view of Hufton et al. ("Sorption Enhanced Reaction Process for Hydrogen Production", AICHE).

Regarding claim 14, Bellows et al. discloses all of the claim limitations as set forth above. Additionally the reference discloses that a conventional processes are used to produce said hydrogen-containing stream gas stream (C2/L28-36) but the reference does not explicitly disclose said water-gas shift reactor being a high temperature water-gas shift reactor.

Hufton et al. teaches a conventional system for preparation of hydrogen-containing stream gas stream wherein said system comprises a first reactor which produces a hydrogen-containing gas stream comprising CO, a water-gas shift reactor and an adsorbent vessel (Fig. 1), wherein the water-gas shift reactor is a high temperature water-gas shift reactor (page 248, column 2, lines 1-2).

It would have been obvious to one having ordinary skill in the art at the time of the invention to use a high temperature water-gas shift reactor in the system of Bellows et al., as taught by Hufton et al., since doing so would amount to nothing more than a use of a known apparatus for its intended use in a known environment to accomplish entirely expected result.

6. Claim 15 is rejected under 35 U.S.C. 103(a) as being unpatentable over Bellows et al. (USP 5,604,047) in view of Smith et al. (USP 3,699,218) or Meyer (USP 3,011,589).

Regarding claim 15, Bellows et al. discloses all of the claim limitations as set forth above and, additionally the reference discloses a preferred temperature range for adsorption (C3/L60-

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C4/L4). While the reference does not explicitly disclose said adsorbent being adapted to adsorb carbon monoxide at low temperatures and to desorb carbon monoxide at high temperatures, it was known in the art at the time of the invention that low temperature improves adsorption of carbon monoxide, similarly as high pressure, see Smith et al. (C5/L38-C6/L19) or Meyer (C6/L59-64 and C7/L25-28). Recitation of well known properties of adsorbent does not confer patentability to the claims.

7. Claims 2-3, 8, 14-15 and 46-47 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hufton et al. ("Sorption Enhanced Reaction Process for Hydrogen Production", AICHE) in view of Meyer (USP 3,011,589).

Regarding claim(s) 46-47, 8 and 15, Hufton et al. disclose(s) system comprising:

- a first reactor which produces a hydrogen-containing stream gas stream comprising CO (Fig. 1);
- a second reactor, which is a water-gas shift reactor disposed downstream of said first reactor, said water-gas shift reactor (Fig. 1); and
- a vessel downstream of said water-gas shift reactor, said vessel housing a second adsorbent adapted to adsorb the carbon monoxide (Fig. 1).

The reference does not explicitly disclose a first adsorbent adapted to adsorb the carbon monoxide in the second reactor.

Meyer teaches a system for production of hydrogen comprising a second adsorbent (C6/L5-14) adapted to remove carbon monoxide from gas comprising hydrogen and carbon monoxide, the system further comprising a first adsorbent (C4/L68-75) adapted to remove carbon monoxide, upstream said second adsorbent. Wherein at least one of the first and second adsorbent is selected from the group consisting of 5A zeolite, 13X zeolite, and mixtures thereof (C4/L68-75 and C6/L5-14) and wherein the first adsorbent is adapted to adsorb carbon monoxide at low temperatures and is

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adapted to desorb carbon monoxide at high temperatures (C6/L59-64 and C7/L25-28). Use of the initial (first) adsorbent improves economy and efficiency of the system (C4/L33-45).

It would have been obvious to one having ordinary skill in the art at the time of the invention to include an initial adsorbent adapted to remove carbon monoxide from gas comprising hydrogen and carbon monoxide in the system of Hufton et al., as taught by Meyer, for the purpose of improving system economy and efficiency.

Further, as the instant specification is silent to unexpected results, it would have been prima facie obvious for one having ordinary skill in the art to combine the water gas shift reactor and the initial adsorber of Hufton et al. in view of Meyer, with the reasonable expectation of eliminating the need for a separate vessels. One of ordinary skill in the art would have been motivated to forgo use of separate vessels, by combining their function and benefit, where doing so is technically feasible and would reduce cost. See *In re Thompson*, 545 F.2d 1290, 1229, 188 USPQ 365, 367 (CCPA 1976). Therefore, one of ordinary skill in the art would have expected that the combination of water-shift reactor and the adsorbent vessel would avoid costs associated with the installation and operation of two vessels. *In re Thompson*, 545 F.2d 1290, 1294, 192 USPQ 275, 277 (CCPA 1976); *In re Clinton*, 527 F.2d 1226, 1229, 188 USPQ 365, 367 (CCPA 1976). Further, the examiner notes that said modification would merely involve making elements integral, and it has been held that making elements integral is generally recognized as being within the level of ordinary skill in the art. *In re Larson*, 340 F.2d 965, 968, 144 USPQ 347, 349 (CCPA 1965).

Regarding claims 2-3 and 14 Hufton et al. in view of Meyer disclose all of the claim limitations as set forth above. Additionally Hufton et al. discloses the system wherein:

- wherein the vessel is a pressure swing adsorber (Fig. 1);
- wherein the pressure swing adsorber comprises multiple staged fixed beds (page 248, column 2,

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lines 19-20); and

- wherein the water gas shift reactor is a high temperature water gas shift reactor (page 248, column 2, lines 1-2).

8. Claim(s) 4-7, 16-18, 20-23, 25-26 and 29-30 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hufton et al. ("Sorption Enhanced Reaction Process for Hydrogen Production", AIChE) in view of Meyer (USP 3,011,589) and further in view of Keefer et al. (WO 00/16425).

Regarding claim(s) 4, Hufton et al. in view of Meyer disclose all of the claim limitations as set forth above, but the references do not explicitly disclose said pressure swing adsorber comprising a rotating vessel, nor do the references disclose the specifics of said vessel.

Keefer et al. teaches a pressure swing adsorption system for hydrogen purification comprising rotating vessel (P25/L10-20 and P27/L31-P28/L8). The rotating pressure swing adsorption system is more efficient than conventional systems (P3/L1-26). Additionally said rotating pressure swing adsorption system is less complicated and less expensive to install and it requires less inventory of adsorbent (P3/L1-26).

It would have been obvious to one having ordinary skill in the art at the time the invention was made to use the rotating PSA vessel of Keefer et al. in the system of Hufton et al. in view of Meyer, for the purpose of increasing system efficiency and lowering capital and operation costs.

Regarding claim(s) 5-6 and 16-17 Hufton et al. in view of Meyer and further in view of Keefer et al. disclose(s) all of the claim limitations as set forth above. Additionally, Keefer et al. discloses the system wherein:

- the rotating vessel comprises an adsorption region, a depressurization region, a purge region and a pressurization region (Fig. 1-6 and P7/L20-P15/L4);
- the rotating vessel comprises two fixed valve faces (P9/L2-9);

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- the system further comprising an expander downstream of the vessel, wherein the expander provides a purge gas to be fed back into the vessel (Fig. 1-4); and
- a fuel cell stack having an anode exhaust, the fuel cell stack disposed between the vessel and the expander, and wherein the expander expands the anode exhaust, the expanded anode exhaust providing the purge gas to be fed back into the vessel (P25/L10-27 and P27/L21-P28/L8).

Regarding claim(s) 7 Hufton et al. in view of Meyer disclose all of the claim limitations as set forth above. While Hufton et al. does not disclose any specific use for produced hydrogen gas, it was well known in the art at the time of the invention that fuel-cells need hydrogen for production of electricity, as evidenced by Keefer et al. (P1/L4-P2/L32). Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use hydrogen produced in the system of Hufton et al. in view of Meyer in a fuel cell system, as doing so would amount to nothing more than use of a known material for its intended use in a known environment to accomplish entirely expected result.

Regarding claim 18, Hufton et al. in view of Meyer and further in view of Keefer et al. disclose all of the claim limitations as set forth above. Additionally, Keefer et al. discloses the system wherein the expander is adapted to provide electrical power for driving the rotating vessel (P11/L6-19, P13/L5-7, P16/L30-P17/L12, P22/L31-P23-9, P25/L10-27 and P27/L21-P28/L8). While the references do not disclose said expander being an isothermal expander, as isothermal expanders are conventional, using an isothermal expander in the system of Hufton et al. in view of Meyer and further in view of Keefer et al. would amount to nothing more than a use of a known apparatus for its intended use in a known environment to accomplish entirely expected result.

Regarding claim(s) 29 Hufton et al. in view of Meyer and further in view of Keefer et al. disclose(s) all of the claim limitations as set forth above. Additionally, all references disclose the

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system wherein the preferential oxidizer is eliminated from the system (Fig. 1 of Hufton et al. and Meyer and Fig. 12-13 of Keefer et al.).

Regarding claim(s) 20 and 25-26, Hufton et al. disclose(s) system comprising:

- a first reactor which produces a hydrogen-containing stream gas stream comprising CO (Fig. 1);
- a second reactor, which is a water-gas shift reactor disposed downstream of said first reactor, said water-gas shift reactor (Fig. 1); and
- a vessel downstream of said water-gas shift reactor, said vessel housing a first adsorbent adapted to adsorb the carbon monoxide (Fig. 1);
- wherein said vessel is a pressure swing adsorber (Fig. 1).

The reference does not explicitly disclose a second adsorbent adapted to adsorb the carbon monoxide in the second reactor. Neither does it disclose the pressure swing adsorber comprising a rotating vessel, nor does the reference disclose the specifics of said vessel.

Meyer teaches a system for production of hydrogen comprising a first adsorbent (C6/L5-14) adapted to remove carbon monoxide from gas comprising hydrogen and carbon monoxide, the system further comprising a second adsorbent (C4/L68-75) adapted to remove carbon monoxide, upstream said first adsorbent. Wherein the first adsorbent is further adapted to adsorb at least one of carbon dioxide and water from the hydrogen gas stream (C4/L68-75 and C6/L5-14). Wherein the first adsorbent is selected from the group consisting of 5A zeolite, 13X zeolite, and mixtures thereof (C4/L68-75 and C6/L5-14) and wherein the second adsorbent is adapted to adsorb carbon monoxide at low temperatures and is adapted to desorb carbon monoxide at high temperatures (C6/L59-64 and C7/L25-28). Use of the initial (second) adsorbent improves economy and efficiency of the system (C4/L33-45).

It would have been obvious to one having ordinary skill in the art at the time of the

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invention to include an initial adsorbent adapted to remove carbon monoxide from gas comprising hydrogen and carbon monoxide in the system of Hufton et al., as taught by Meyer, for the purpose of improving system economy and efficiency.

Further, as the instant specification is silent to unexpected results, it would have been prima facie obvious for one having ordinary skill in the art to combine the water gas shift reactor and the initial adsorber of Hufton et al. in view of Meyer, with the reasonable expectation of eliminating the need for a separate vessels. One of ordinary skill in the art would have been motivated to forgo use of separate vessels, by combining their function and benefit, where doing so is technically feasible and would reduce cost. See *In re Thompson*, 545 F.2d 1290, 1229, 188 USPQ 365, 367 (CCPA 1976). Therefore, one of ordinary skill in the art would have expected that the combination of water-shift reactor and the adsorbent vessel would avoid costs associated with the installation and operation of two vessels. *In re Thompson*, 545 F.2d 1290, 1294, 192 USPQ 275, 277 (CCPA 1976); *In re Clinton*, 527 F.2d 1226, 1229, 188 USPQ 365, 367 (CCPA 1976). Further, the examiner notes that said modification would merely involve making elements integral, and it has been held that making elements integral is generally recognized as being within the level of ordinary skill in the art. *In re Larson*, 340 F.2d 965, 968, 144 USPQ 347, 349 (CCPA 1965).

Keefer et al. teaches a pressure swing adsorption system for hydrogen purification comprising rotating vessel (P25/L10-20 and P27/L31-P28/L8). Wherein the rotating vessel comprises an adsorption region, a depressurization region, a purge region and a pressurization region (Fig. 1-6 and P7/L20-P15/L4); and comprises two fixed valve faces (P9/L2-9). The rotating pressure swing adsorption system is more efficient than conventional systems (P3/L1-26). Additionally said rotating pressure swing adsorption system is less complicated and less expensive to install and it requires less inventory of adsorbent (P3/L1-26).

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It would have been obvious to one having ordinary skill in the art at the time the invention was made to use the rotating PSA vessel of Keefer et al. in the system of Hufton et al. in view of Meyer, for the purpose of increasing system efficiency and lowering capital and operation costs.

Regarding claim(s) 21-22 Hufton et al. in view of Meyer and further in view of Keefer et al. disclose(s) all of the claim limitations as set forth above. Additionally, Keefer et al. discloses the system wherein:

- the system further comprising an expander downstream of the vessel, wherein the expander provides a purge gas to be fed back into the vessel (Fig. 1-4); and
- a fuel cell stack having an anode exhaust, the fuel cell stack disposed between the vessel and the expander, and wherein the expander expands the anode exhaust, the expanded anode exhaust providing the purge gas to be fed back into the vessel (P25/L10-27 and P27/L21-P28/L8).

Regarding claim 23, Hufton et al. in view of Meyer and further in view of Keefer et al. disclose all of the claim limitations as set forth above. Additionally, Keefer et al. discloses the system wherein the expander is adapted to provide electrical power for driving the rotating vessel (P11/L6-19, P13/L5-7, P16/L30-P17/L12, P22/L31-P23-9, P25/L10-27 and P27/L21-P28/L8).

While the references do not disclose said expander being an isothermal expander, as isothermal expanders are conventional, using an isothermal expander in the system of Hufton et al. in view of Meyer and further in view of Keefer et al. would amount to nothing more than a use of a known apparatus for its intended use in a known environment to accomplish entirely expected result.

Regarding claim(s) 30 Hufton et al. in view of Meyer and further in view of Keefer et al. disclose(s) all of the claim limitations as set forth above. Additionally, all references disclose discloses the system wherein the preferential oxidizer is eliminated from the system (Fig. 1 of Hufton et al. and Meyer and Fig. 12-13 of Keefer et al.).

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9. Claim(s) 9-11 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hufton et al. ("Sorption Enhanced Reaction Process for Hydrogen Production", AICHE), in view of Meyer (USP 3,011,589) and further in view of Kirshnamurthy (USP 5,096,470).

Regarding claim(s) 9, Hufton et al. in view of Meyer disclose(s) all of the claim limitations as set forth above. Additionally Hufton et al. is not limited to any specific adsorbents which can be used in PSA vessel. As adsorbents selected from the group consisting of oxides or salts of copper impregnated or exchanged on activated carbon, alumina and zeolites, oxides or salts of silver impregnated or exchanged on activated carbon, alumina and zeolites, oxides or salts of tin impregnated or exchanged on activated carbon, alumina and zeolites, and mixtures thereof, were known to be useful in the PSA systems for separation of hydrogen from gas streams containing hydrogen and CO, as evidenced by Kirshnamurthy (C7/L14-C8/L2), and further, as the instant specification is silent to unexpected results, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use any adsorbent disclosed by Kirshnamurthy in the PSA system of Hufton et al., as doing so would amount to nothing more than to use of a known material for its intended use in a known environment to accomplish entirely expected result.

Regarding claim(s) 10-11, Hufton et al. in view of Meyer disclose(s) all of the claim limitations as set forth above, but it does not explicitly disclose a desiccant material selected from the group consisting of zeolite molecular sieves, activated alumina, silica gels, and mixtures thereof disposed upstream of the CO adsorbent.

Kirshnamurthy teaches that is known in the art to pass the feed to the PSA adsorber through a layer of desiccant selected from the group consisting of zeolite molecular sieves, activated alumina, silica gels, and mixtures thereof for the purpose of removing water from said feed..

It would have been obvious to one having ordinary skill in the art at the time the invention

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was made to use layer of desiccant, as taught by of Kirshnamurthy upstream of the adsorbent of Hufton et al. in view of Meyer, for the purpose of improving system operation by removing water from feed to said adsorbent.

10. Claim(s) 27-28 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hufton et al. ("Sorption Enhanced Reaction Process for Hydrogen Production", AICHE), in view of Meyer (USP 3,011,589), in view of Keefer et al. (WO 00/16425) and further in view of Kirshnamurthy (USP 5,096,470).

Regarding claim(s) 27, Hufton et al. in view of Meyer and further in view of Keefer et al. disclose(s) all of the claim limitations as set forth above. Additionally Hufton et al. is not limited to any specific adsorbents which can be used in PSA vessel. As adsorbents selected from the group consisting of oxides or salts of copper impregnated or exchanged on activated carbon, alumina and zeolites, oxides or salts of silver impregnated or exchanged on activated carbon, alumina and zeolites, oxides or salts of tin impregnated or exchanged on activated carbon, alumina and zeolites, and mixtures thereof, were known to be useful in the PSA systems for separation of hydrogen from gas streams containing hydrogen and CO, as evidenced by Kirshnamurthy (C7/L14-C8/L2), and further, as the instant specification is silent to unexpected results, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use any adsorbent disclosed by Kirshnamurthy in the PSA system of Hufton et al., as doing so would amount to nothing more than to use of a known material for its intended use in a known environment to accomplish entirely expected result.

Regarding claim(s) 28, Hufton et al. in view of Meyer and further in view of Keefer et al. disclose(s) all of the claim limitations as set forth above, but it does not explicitly disclose a desiccant material selected from the group consisting of zeolite molecular sieves, activated alumina,

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silica gels, and mixtures thereof disposed upstream of the CO adsorbent.

Kirshnamurthy teaches that is known in the art to pass the feed to the PSA adsorber through a layer of desiccant selected from the group consisting of zeolite molecular sieves, activated alumina, silica gels, and mixtures thereof for the purpose of removing water from said feed.

It would have been obvious to one having ordinary skill in the art at the time the invention was made to use layer of desiccant, as taught by of Kirshnamurthy upstream of the adsorbent of Hufton et al. in view of Meyer and further in view of Keefer et al., for the purpose of improving system operation by removing water from feed to said adsorbent.

11. Claim(s) 2-8, 14-18, 20-23, 25-26, 29-30 and 46-47 are rejected under 35 U.S.C. 103(a) as being unpatentable over Keefer et al. (WO 00/16425) in view of Meyer (USP 3,011,589).

Regarding claim(s) 46-47, 8 and 15, Keefer et al. disclose(s) system comprising:

- a first reactor (445 and 544) which produces a hydrogen-rich gas stream comprising CO; and
- a second reactor (448 and 548) which is a water gas shift reactor disposed downstream of said first reactor (445 and 544);
- a vessel (432 and 532) downstream of said water-gas shift reactor (448 and 548), said vessel housing a second adsorbent adapted to adsorb the carbon monoxide.

The reference does not explicitly disclose a first adsorbent adapted to adsorb the carbon monoxide in the second reactor.

Meyer teaches a system for production of hydrogen comprising a second adsorbent (C6/L5-14) adapted to remove carbon monoxide from gas comprising hydrogen and carbon monoxide, the system further comprising a first adsorbent (C4/L68-75) adapted to remove carbon monoxide, upstream said second adsorbent. Wherein at least one of the first and second adsorbent is selected from the group consisting of 5A zeolite, 13X zeolite, and mixtures thereof (C4/L68-75 and C6/L5-

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14) and wherein the first adsorbent is adapted to adsorb carbon monoxide at low temperatures and is adapted to desorb carbon monoxide at high temperatures (C6/L59-64 and C7/L25-28). Use of initial (first) adsorbent improves economy and efficiency of the system (C4/L33-45).

It would have been obvious to one having ordinary skill in the art at the time of the invention to include an initial adsorbent adapted to remove carbon monoxide from gas comprising hydrogen and carbon monoxide in the system of Keefer et al., as taught by Meyer, for the purpose of improving system economy and efficiency.

Further, as the instant specification is silent to unexpected results, it would have been prima facie obvious for one having ordinary skill in the art to combine the water gas shift reactor and the initial adsorber of Keefer et al. in view of Meyer, with the reasonable expectation of eliminating the need for a separate vessels. One of ordinary skill in the art would have been motivated to forgo use of separate vessels, by combining their function and benefit, where doing so is technically feasible and would reduce cost. See *In re Thompson*, 545 F.2d 1290, 1229, 188 USPQ 365, 367 (CCPA 1976). Therefore, one of ordinary skill in the art would have expected that the combination of water-shift reactor and the adsorbent vessel would avoid costs associated with the installation and operation of two vessels. *In re Thompson*, 545 F.2d 1290, 1294, 192 USPQ 275, 277 (CCPA 1976); *In re Clinton*, 527 F.2d 1226, 1229, 188 USPQ 365, 367 (CCPA 1976). Further, the examiner notes that said modification would merely involve making elements integral, and it has been held that making elements integral is generally recognized as being within the level of ordinary skill in the art. *In re Larson*, 340 F.2d 965, 968, 144 USPQ 347, 349 (CCPA 1965).

Regarding claims 2-7, 14, 16-17 and 29 Keefer et al. in view of Meyer disclose(s) all of the claims limitations as set forth above. Additionally Keefer et al. discloses the system wherein:

- wherein the vessel is a pressure swing adsorber (P25/L10-20 and P27/L31-P28/L8);

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- wherein the pressure swing adsorber comprises multiple, staged fixed beds (Fig. 1-6);
- wherein the pressure swing adsorber is a rotating vessel (P25/L10-20 and P27/L31-P28/L8);
- wherein the rotating vessel comprises an adsorption region, a depressurization region, a purge region and a pressurization region (Fig. 1-6 and P7/L20-P15/L4);
- wherein the rotating vessel comprises two fixed valve faces (P9/L2-9);
- wherein the system is a fuel cell system (abstract);
- wherein the water gas shift reactor is a high temperature water gas shift reactor (P25/L28-P26/L31);
- the system further comprising an expander downstream of the vessel, wherein the expander provides a purge gas to be fed back into the vessel (Fig. 1-4); and
- a fuel cell stack having an anode exhaust, the fuel cell stack disposed between the vessel and the expander, and wherein the expander expands the anode exhaust, the expanded anode exhaust providing the purge gas to be fed back into the vessel (P25/L10-27 and P27/L21-P28/L8);
- wherein the preferential oxidizer is eliminated from the system (Fig. 12-13).

Regarding claim 18, Keefer et al. in view of Meyer disclose all of the claim limitations as set forth above. Additionally, the reference discloses the system wherein the expander is adapted to provide electrical power for driving the rotating vessel (P11/L6-19, P13/L5-7, P16/L30-P17/L12, P22/L31-P23-9, P25/L10-27 and P27/L21-P28/L8). While the references do not disclose said expander being an isothermal expander, as isothermal expanders are conventional, using an isothermal expander in the system of Keefer et al. in view of Meyer would amount to nothing more than a use of a known apparatus for its intended use in a known environment to accomplish entirely expected result.

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Regarding claim(s) 20 and 25-26, Keefer et al. disclose(s) system comprising:

- a first reactor (445 and 544) which produces a hydrogen-rich gas stream comprising CO; and
- a second reactor (448 and 548) which is a water gas shift reactor disposed downstream of said first reactor (445 and 544);
- a rotating vessel (432 and 532) downstream of said water-gas shift reactor (448 and 548), said rotating vessel housing a second adsorbent adapted to adsorb the carbon monoxide (P25/L10-20 and P27/L31-P28/L8);
- wherein the rotating vessel is a pressure swing adsorber comprising an adsorption region, a depressurization region, a purge region and a pressurization region (Fig. 1-6 and P7/L20-P15/L4); and comparing two fixed valve faces (P9/L2-9).

The reference does not explicitly disclose a second adsorbent adapted to adsorb the carbon monoxide in the second reactor.

Meyer teaches a system for production of hydrogen comprising a first adsorbent (C6/L5-14) adapted to remove carbon monoxide from gas comprising hydrogen and carbon monoxide, the system further comprising a second adsorbent (C4/L68-75) adapted to remove carbon monoxide, upstream said first adsorbent. Wherein the first adsorbent is further adapted to adsorb at least one of carbon dioxide and water from the hydrogen gas stream (C4/L68-75 and C6/L5-14). Wherein the first adsorbent is selected from the group consisting of 5A zeolite, 13X zeolite, and mixtures thereof (C4/L68-75 and C6/L5-14) and wherein the second adsorbent is adapted to adsorb carbon monoxide at low temperatures and is adapted to desorb carbon monoxide at high temperatures (C6/L59-64 and C7/L25-28). Use of the initial (second) adsorbent improves economy and efficiency of the system (C4/L33-45).

It would have been obvious to one having ordinary skill in the art at the time of the

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invention to include an initial adsorbent adapted to remove carbon monoxide from gas comprising hydrogen and carbon monoxide in the system of Keefer et al., as taught by Meyer, for the purpose of improving system economy and efficiency.

Further, as the instant specification is silent to unexpected results, it would have been prima facie obvious for one having ordinary skill in the art to combine the water gas shift reactor and the initial adsorber of Keefer et al. in view of Meyer, with the reasonable expectation of eliminating the need for a separate vessels. One of ordinary skill in the art would have been motivated to forgo use of separate vessels, by combining their function and benefit, where doing so is technically feasible and would reduce cost. See *In re Thompson*, 545 F.2d 1290, 1229, 188 USPQ 365, 367 (CCPA 1976). Therefore, one of ordinary skill in the art would have expected that the combination of water-shift reactor and the adsorbent vessel would avoid costs associated with the installation and operation of two vessels. *In re Thompson*, 545 F.2d 1290, 1294, 192 USPQ 275, 277 (CCPA 1976); *In re Clinton*, 527 F.2d 1226, 1229, 188 USPQ 365, 367 (CCPA 1976). Further, the examiner notes that said modification would merely involve making elements integral, and it has been held that making elements integral is generally recognized as being within the level of ordinary skill in the art. *In re Larson*, 340 F.2d 965, 968, 144 USPQ 347, 349 (CCPA 1965).

Regarding claim(s) 21-22 Keefer et al. in view of Meyer disclose(s) all of the claim limitations as set forth above. Additionally, Keefer et al. discloses the system wherein:

- the system further comprising an expander downstream of the vessel, wherein the expander provides a purge gas to be fed back into the vessel (Fig. 1-4); and
- a fuel cell stack having an anode exhaust, the fuel cell stack disposed between the vessel and the expander, and wherein the expander expands the anode exhaust, the expanded anode exhaust providing the purge gas to be fed back into the vessel (P25/L10-27 and P27/L21-P28/L8).

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Regarding claim 23, Keefer et al. in view of Meyer disclose all of the claim limitations as set forth above. Additionally, Keefer et al. discloses the system wherein the expander is adapted to provide electrical power for driving the rotating vessel (P11/L6-19, P13/L5-7, P16/L30-P17/L12, P22/L31-P23-9, P25/L10-27 and P27/L21-P28/L8). While the references do not disclose said expander being an isothermal expander, as isothermal expanders are conventional, using an isothermal expander in the system of Hufton et al. in view of Meyer and further in view of Keefer et al. would amount to nothing more than a use of a known apparatus for its intended use in a known environment to accomplish entirely expected result.

Regarding claim(s) 30 Keefer et al. in view of Meyer disclose(s) all of the claim limitations as set forth above. Additionally, all references disclose discloses the system wherein the preferential oxidizer is eliminated from the system (Fig. 1 of Meyer and Fig. 12-13 of Keefer et al.).

12. Claim(s) 9-11 and 27-28 are rejected under 35 U.S.C. 103(a) as being unpatentable over Keefer et al. (WO 00/16425) in view of Meyer (USP 3,011,589) and further in view of Kirshnamurthy (USP 5,096,470).

Regarding claim(s) 9, Keefer et al. in view of Meyer disclose all of the claim limitations as set forth above. Additionally Keefer et al. discloses that supported zeolite type adsorbent can be used in the vessel (P8/L14-29 and P17/L22-33). As adsorbents selected from the group consisting of oxides or salts of copper impregnated or exchanged on activated carbon, alumina and zeolites, oxides or salts of silver impregnated or exchanged on activated carbon, alumina and zeolites, oxides or salts of tin impregnated or exchanged on activated carbon, alumina and zeolites, and mixtures thereof, were known to be useful in the PSA systems for separation of hydrogen from gas streams containing hydrogen and CO, as evidenced by Kirshnamurthy (C7/L14-C8/L2), and further, as the instant specification is silent to unexpected results, it would have been obvious to one having

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ordinary skill in the art at the time the invention was made to use any adsorbent disclosed by Kirshnamurthy in the PSA system of Keefer et al. in view of Meyer, as doing so would amount to nothing more than to use of a known material for its intended use in a known environment to accomplish entirely expected result.

Regarding claim(s) 10-11, Keefer et al. in view of Meyer disclose all of the claim limitations as set forth above and additionally Keefer et al. discloses that water vapor can deactivate the adsorbents (P24/L5-14), but it does not explicitly disclose a desiccant material selected from the group consisting of zeolite molecular sieves, activated alumina, silica gels, and mixtures thereof disposed upstream of the CO adsorbent.

Kirshnamurthy teaches that is known in the art to pass the feed to the PSA adsorber through a layer of desiccant selected from the group consisting of zeolite molecular sieves, activated alumina, silica gels, and mixtures thereof for the purpose of removing water from said feed..

It would have been obvious to one having ordinary skill in the art at the time the invention was made to use layer of desiccant, as taught by of Kirshnamurthy upstream of the adsorbent of Keefer et al. in view of Meyer, for the purpose of improving system operation by removing water from feed to said adsorbent.

Regarding claim(s) 27, Keefer et al. in view of Meyer disclose(s) all of the claim limitations as set forth above. Additionally Keefer et al. discloses that supported, zeolite type adsorbent can be used in the vessel (P8/L14-29 and P17/L22-33). As adsorbents selected from the group consisting of oxides or salts of copper impregnated or exchanged on activated carbon, alumina and zeolites, oxides or salts of silver impregnated or exchanged on activated carbon, alumina and zeolites, oxides or salts of tin impregnated or exchanged on activated carbon, alumina and zeolites, and mixtures thereof, were known to be useful in the PSA systems for separation of hydrogen from gas streams

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containing hydrogen and CO, as evidenced by Kirshnamurthy (C7/L14-C8/L2), and further, as the instant specification is silent to unexpected results, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use any adsorbent disclosed by Kirshnamurthy in the PSA system of Keefer et al., as doing so would amount to nothing more than to use of a known material for its intended use in a known environment to accomplish entirely expected result.

Regarding claim(s) 28, Keefer et al. in view of Meyer disclose(s) all of the claim limitations as set forth above and additionally Keefer et al. discloses that water vapor can deactivate the adsorbents (P24/L5-14), but it does not explicitly disclose a desiccant material selected from the group consisting of zeolite molecular sieves, activated alumina, silica gels, and mixtures thereof disposed upstream of the CO adsorbent.

Kirshnamurthy teaches that is known in the art to pass the feed to the PSA adsorber through a layer of desiccant selected from the group consisting of zeolite molecular sieves, activated alumina, silica gels, and mixtures thereof for the purpose of removing water from said feed.

It would have been obvious to one having ordinary skill in the art at the time the invention was made to use layer of desiccant, as taught by of Kirshnamurthy upstream of the adsorbent of Keefer et al. in view of Meyer, for the purpose of improving system operation by removing water from feed to said adsorbent.

13. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made

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in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

Double Patenting

14. The nonstatutory double patenting rejection is based on a judicially created doctrine grounded in public policy (a policy reflected in the statute) so as to prevent the unjustified or improper timewise extension of the "right to exclude" granted by a patent and to prevent possible harassment by multiple assignees. See *In re Goodman*, 11 F.3d 1046, 29 USPQ2d 2010 (Fed. Cir. 1993); *In re Longi*, 759 F.2d 887, 225 USPQ 645 (Fed. Cir. 1985); *In re Van Ornum*, 686 F.2d 937, 214 USPQ 761 (CCPA 1982); *In re Vogel*, 422 F.2d 438, 164 USPQ 619 (CCPA 1970); and, *In re Thorington*, 418 F.2d 528, 163 USPQ 644 (CCPA 1969).

A timely filed terminal disclaimer in compliance with 37 CFR 1.321(c) may be used to overcome an actual or provisional rejection based on a nonstatutory double patenting ground provided the conflicting application or patent is shown to be commonly owned with this application. See 37 CFR 1.130(b).

Effective January 1, 1994, a registered attorney or agent of record may sign a terminal disclaimer. A terminal disclaimer signed by the assignee must fully comply with 37 CFR 3.73(b).

15. Claim(s) 2-11, 14-18, 20-23, 25-30 and 46-47 are rejected under the judicially created doctrine of obviousness-type double patenting as being unpatentable over claims 1-34 of U.S. Patent No. 6,692,545 in view of in view of Meyer (USP 3,011,589) and further in view of Kirshnamurthy (USP 5,096,470).

Claims 1-34 of U.S. Patent No. 6,692,545 recite all of the claim limitations as recited in rejected claims, but they do not recite a rotating vessel comprising carbon monoxide adsorbent.

Meyer teaches a system for production of hydrogen comprising a two adsorbent zones (C6/L5-14) adapted to remove carbon monoxide from gas comprising hydrogen and carbon monoxide. Use of two consecutive adsorbents improves economy and efficiency of the system (C4/L33-45).

It would have been obvious to one having ordinary skill in the art at the time of the invention to include a vessel comprising a second adsorbent adapted to remove carbon monoxide from gas comprising hydrogen and carbon monoxide in the system of recited in claims 1-34 of U.S.

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Patent No. 6,692,545, as taught by Meyer, for the purpose of improving system economy and efficiency.

Keefer et al. teaches adsorbent for removing carbon monoxide from hydrogen comprising gas, said adsorbent comprising rotating vessel (P25/L10-20 and P27/L31-P28/L8). Wherein the rotating vessel comprises an adsorption region, a depressurization region, a purge region and a pressurization region (Fig. 1-6 and P7/L20-P15/L4); and comprises two fixed valve faces (P9/L2-9). The rotating pressure swing adsorption system is more efficient than conventional systems (P3/L1-26). Additionally said rotating pressure swing adsorption system is less complicated and less expensive to install and it requires less inventory of adsorbent (P3/L1-26).

It would have been obvious to one having ordinary skill in the art at the time the invention was made to use the rotating PSA vessel of Keefer et al. as the second adsorbent vessel in the system of recited in claims 1-34 of U.S. Patent No. 6,692,545, in view of Meyer, for the purpose of increasing system efficiency and lowering capital and operation costs.

Response to Arguments

16. Applicant's arguments filed on 17 June 2004 have been fully considered but they are not persuasive.

17. Applicant's arguments with respect to rejection of claims under 35 U.S.C. § 112, second paragraph are not clear, as the examiner has not rejected any claims under 35 U.S.C. § 112, second paragraph.

18. The applicant argues that the specification on page 10, lines 11-21 describes an embodiment wherein the adsorbent zone is located downstream from the water gas shift zone. The examiner would like to reiterate that the recitation of apparatus comprising a second reactor which is a water gas shift reactor, wherein the water gas shift reactor includes an adsorbent adapted to adsorb carbon

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monoxide is not enabled by the specification because it encompasses systems wherein the adsorbent included in the water gas shift reactor is placed after the water gas shift catalyst. The adsorbent zone located downstream from the water gas shift zone, as described on page 10, lines 11-21, is not included in the water gas shift zone (reference number 2), but rather it is located in a separate vessel (reference number 3). The only adsorbent that is included in the water gas shift zone, as disclosed on page 10, lines 11-21, is added into or before the water gas shift catalyst.

19. The mere recognition that system wherein carbon monoxide adsorbent is included in the water gas shift zone improves start-up performance, overall system efficiency and fuel consumption does not impart patentability to the claimed subject matter. See *Ex parte Obiaya*, 227 USPQ 58 (Bd. Pat. App. & Int. 1985), *aff'd*, 795 F.2d 1017 (Fed. Cir. 1986) ("[t]he fact that appellant has recognized another advantage which would flow naturally from following the suggestion of the prior art cannot be the basis for patentability when the differences would otherwise have been obvious"). Applicant has not relied on any factual evidence to demonstrate that argued advantages are unexpected.

20. Applicant's statement that "It is admitted in the Office Action that Bellows does not show first and second sequentially operated absorbers" is not clear, because said Office Action does not include such statement. Further, claims 7, 14-15, 29 and 46, which are rejected over Bellows et al. either by itself or in combination with other references do not recite "first and second sequentially operated absorbers". Although the claims are interpreted in light of the specification, limitations from the specification are not read into the claims. See *In re Van Geuns*, 988 F.2d 1181, 26 USPQ2d 1057 (Fed. Cir. 1993).

21. The applicant argues that Bellows does not show or suggest combining water gas shift reactor and adsorbent. In response the examiner would like to point out that once a reference

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discloses water gas shift reactor and adsorbent it would have been prima facie obvious for one having ordinary skill in the art to combine the water gas shift reactor and the reactor containing the adsorbent into one vessel, with the reasonable expectation of eliminating the need for a two separate vessels. One of ordinary skill in the art would have been motivated to forgo use of separate vessels, by combining their function and benefit, where doing so is technically feasible and would reduce cost. See *In re Thompson*, 545 F.2d 1290, 1229, 188 USPQ 365, 367 (CCPA 1976). Therefore, one of ordinary skill in the art would have expected that the combination of water-shift reactor and the adsorbent vessel would avoid costs associated with the installation and operation of two vessels. *In re Thompson*, 545 F.2d 1290, 1294, 192 USPQ 275, 277 (CCPA 1976); *In re Clinton*, 527 F.2d 1226, 1229, 188 USPQ 365, 367 (CCPA 1976). Further, the examiner notes that said modification would merely involve making elements integral, and it has been held that making elements integral is generally recognized as being within the level of ordinary skill in the art. *In re Larson*, 340 F.2d 965, 968, 144 USPQ 347, 349 (CCPA 1965).

22. Applicant's statement that "It is not understood how claims 29 and 7, which respectively depend on claim 20 and 46, are rejected on the basis of Bellows" is not clear, because neither claim 29 nor claim 7 depends on claim 20. The grounds of rejection of claims 46, 7 and 9, based on Bellows et al., were set forth in paragraph 6 of prior Office Action.

23. In response to applicant's argument that Meyer is not able to be combined with any of the art of record, the examiner notes that the test for obviousness is not whether the features of a secondary reference may be bodily incorporated into the structure of the primary reference; nor is it that the claimed invention must be expressly suggested in any one or all of the references. Rather, the test is what the combined teachings of the references would have suggested to those of ordinary skill in the art. See *In re Keller*, 642 F.2d 413, 208 USPQ 871 (CCPA 1981). In the instant case Meyer is

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relied upon to show well known properties of carbon monoxide adsorbent (C6/L59-64 and C7/L25-28) or to teach a system for removal of carbon monoxide from stream containing carbon monoxide and hydrogen using two carbon monoxide adsorbents, in sequence (C6/L5-14). Said teachings can be combined with teachings of other references which are concerned with removing of carbon monoxide from stream comprising carbon monoxide and hydrogen.

24. In response to applicant's argument that Meyer is nonanalogous art, it has been held that a prior art reference must either be in the field of applicant's endeavor or, if not, then be reasonably pertinent to the particular problem with which the applicant was concerned, in order to be relied upon as a basis for rejection of the claimed invention. See *In re Oetiker*, 977 F.2d 1443, 24 USPQ2d 1443 (Fed. Cir. 1992). In this case, Meyer, similarly to applicant's invention and to other prior art references, is concerned with providing pure hydrogen stream by removing carbon monoxide.

25. Applicant argues that Hufton et al. teaches away system shown in Fig. 1, because the preferred system is shown in Fig. 9. This is not found persuasive. While Hufton et al. may describe some systems as more desirable than others, such disclosure does not take away from the fact that all disclosed systems, were known to one having ordinary skill in the art at the time of the invention. In fact the system shown in Fig. 1 of Hufton et al. is referred to as a "conventional" system for hydrogen production.

26. The applicant argues that there is no suggestion in Smith et al. to combine the water gas shift reactor and adsorber functions together. This is not found persuasive because the examiner has not relied upon Smith et al. to teach combination of the water gas shift reactor and adsorber functions together. In the instant case Smith et al. is relied upon to show well known properties of carbon monoxide adsorbent (C5/L38-C6/L9).

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27. The applicant argues that Kirshnamurthy does not address how the hydrogen/carbon monoxide stream is produced in the first place. This is not found persuasive because the examiner has not relied upon Kirshnamurthy to teach how the hydrogen/carbon monoxide stream is produced. The examiner has relied upon either Bellows et al., Hufton et al. or Keefer et al., as set forth above, to teach how the hydrogen/carbon monoxide stream is produced.

28. The applicant argues that Hufton et al. is not combinable with Keefer et al., because Fig. 9 of Hufton et al. lacks the components of Keefer et al. This is not found persuasive. While Hufton et al. may describe some systems as more desirable than others, such disclosure does not take away from the fact that all disclosed systems, were known to one having ordinary skill in the art at the time of the invention. In fact the system shown in Fig. 1 of Hufton et al. is referred to as a "conventional" system for hydrogen production. The examiner would like to reiterate that the rejections are based on system disclosed in Fig. 1 of Hufton et al., and not on Fig. 9. Additionally, the examiner notes that the test for obviousness is not whether the features of a secondary reference may be bodily incorporated into the structure of the primary reference; nor is it that the claimed invention must be expressly suggested in any one or all of the references. Rather, the test is what the combined teachings of the references would have suggested to those of ordinary skill in the art. See *In re Keller*, 642 F.2d 413, 208 USPQ 871 (CCPA 1981).

29. Regarding rejection of Bellows et al. in view of Hufton et al. the applicant argues that said references can not be combined because while Bellows et al. shows a steam reformer and a separate water gas shift reactor, Hufton et al. teaches combination of steam reformer and adsorber with no water gas shift reactor. This is not found persuasive, because Hufton et al., in fig. 1, clearly teaches a steam reformer and a separate water gas shift reactor.

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30. Applicant's arguments that Keefer et al. and Meyer are not combinable because Keefer et al. shows system elements similar to Fig. 1 of Hufton et al. were considered, but are not clear.

31. Regarding the double patenting rejection, the applicant argues that the system of patent 6,692,545 includes apparatus for removing carbon monoxide using an adsorbent which adsorbs carbon dioxide, and said feature is not recited in the independent claims of the instant application. This is not found persuasive. Instant claims are rejected over claims 1-34 of U.S. Patent No. 6,692,545 in view of Meyer and further in view of Kirshnamurthy because all of the limitations recited in the instant claims are recited in claims 1-34 of U.S. Patent No. 6,692,545 in view of said references, and not vice versa. Instant claims do not exclude systems which include carbon dioxide adsorbent in the water gas shift reactor in addition to carbon monoxide adsorbent.

Conclusion

32. **THIS ACTION IS MADE FINAL.** Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.


33. Any inquiry concerning this communication or earlier communications from the examiner should be directed to examiner Basia Ridley, whose telephone number is (571) 272-1453.

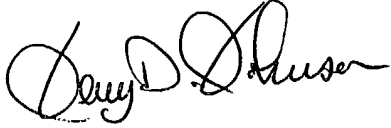
If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Glenn Caldarola, can be reached on (571) 272-1444.

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The fax phone number for the organization where this application or proceeding is assigned is (703) 872-9306.

Technical Center 1700 General Information Telephone No. is (571) 272-1700. Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Questions on access to the Private PAIR system should be directed to the Electronic Business Center (EBC) at (866) 217-9197 (toll-free).


Basia Ridley
Examiner
Art Unit 1764


JERRY D. JOHNSON
PRIMARY EXAMINER
GROUP 1100

BR
September 17, 2004